

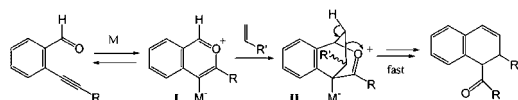
Stereocontrolled Synthesis of Complicated Oxacyclic Compounds via Platinum-Catalyzed [4 + 2]-Cycloadditions and Annulations of Enynals with Allylic Alcohols

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Metal-catalyzed cycloaddition reactions are powerful tools in organic synthesis to access complex molecular frameworks.¹ In the presence of electrophilic metals, common 2-alkynylbenzaldehydes form metal-benzopyrylium intermediates **I** that undergo [4 + 2] cycloaddition with alkenes to form cycloadducts **II**.^{2,3} These species suffer kinetic instability and undergo rapid rearrangement to give 1,2-dihydronaphalene or naphthalene products. Metal-free [4 + 2]-cycloadducts resembling **II** were also too unstable to isolate.^{4,5} The high reactivity of this species allows its interception with enynals of special types that however only gave distinct [3 + 2]-cycloadducts.⁶ We sought new approaches for elaboration of intermediates **II** because of their skeletal complexity. Herein, we report the synthesis of complex oxacyclic compounds through diversified interceptions of intermediate **II** with 2-substituted allylic alcohol; this work first reveals the high diastereoselectivity of the benzopyrylium-alkene [4 + 2] cycloadditions.



As shown in Table 1, treatment of 2-alkynylbenzaldehydes (**1a**) with 2-phenylallylic alcohol (**2a**) with PtCl₂/CO catalyst (10 mol %) in hot toluene (70 °C, 4 h) provided tetracyclic ketal **3a** and oxacyclic ketone **4a** in 35% and 7%, respectively; both species were obtained as a single stereoisomer. Gratifyingly, the use of 1,4-dioxane solvent greatly improved the yields of **3a** and **4a** to 62% and 12%, respectively. Other solvents gave the following yields: benzene (40% **3a**, 8% **4a**), THF (52% **3a**, 12% **4a**) and MeCN (0% for **3a** and **4a**). Commonly used gold catalysts AuCl, AuCl₃, and PPh₃AuCl/AgSbF₆ in dioxane at 23 °C led to complete consumption of starting **1a** with low yields of desired **3a** (<14%). The molecular structures of **3a** and **4a** are inferred through X-ray diffraction studies⁷ of their related analogues **3i** and **6h**.

We examined the cycloaddition reactions of various aldehyde substrates **1a–d** with allylic alcohols **2b–d** to assess the generality of tetracyclic ketal synthesis; the results are depicted in Table 2. All reactions were performed in 1,4-dioxane at 80 °C (2 h) except entry 3 that employed siloxy **2d** in wet and hot toluene. Entries 1–3 showed the suitability of this cycloaddition to allylic alcohols **2b–2c** or siloxy species **2d** (R² = Me, *n*-Bu, TMSCH₂), providing ketals **3b–3d** in 61–72% yields. The cycloadditions are also extendible to various 2-alkynylbenzaldehydes **1b** and **1c**, producing ketals **3e–3h** efficiently (69–78% yields) except for **3i** that was obtained in 41% yield. The X-ray structure of ketal **3i** is provided in Supporting Information.

Table 3 depicts a notable change of chemoselectivity when we examined this platinum catalysis on nonaromatic enynals **5b–f**. On the basis of catalyst screening,⁸ PtCl₂/CO is less efficient as PtCl₂/AgOTf (10/20 mol %), which implements a new annulation of these enynals with allylic alcohols **2a–c** in toluene at 23 °C (16 h). Particularly notable is the formation of a single stereoisomer for oxacyclic products **6a–g** and **6i** despite their molecular complexity;

Table 1. Stereocontrolled Formation of Stable [4 + 2]-Cycloadducts

entry	catalyst (mol %) ^a	condition	product (yields) ^b
1	PtCl ₂ (10)/CO	toluene (80 °C, 4 h)	3a (35%), 4a (7%)
2	PtCl ₂ (10)/CO	1,4-dioxane (80 °C, 2 h)	3a (62%), 4a (12%)
3	AuCl (5)	1,4-dioxane (23 °C, 20 min)	3a (11%)
4	AuCl ₃ (5)	1,4-dioxane (23 °C, 20 min)	3a (14%)
5	AuCIPPh ₃ (%) / AgSbF ₆ (5)	1,4-dioxane (23 °C, 20 min)	3a (12%), 4a (2%)

^a [1a] = 0.17 M, **2a** (2.5 equiv). ^b Product yields are reported after silica column chromatography.

Table 2. Stereocontrolled [4 + 2] Cycloadditions of 2-Alkynylbenzaldehydes with Allylic Alcohols

(1) 1a (R ¹ = <i>n</i> -Bu) ^a	2b (R ² = Me) ^b	3b (72%)	—
(2) 1a (R ¹ = <i>n</i> -Bu)	2c (R ² = <i>n</i> -Bu)	3c (64%)	—
(3) 1a (R ¹ = <i>n</i> -Bu)	2d (R ² = TMSCH ₂)	3d (61%); 3b (4%)	—
(4) 1b (R ¹ = Me)	2b (R ² = Me)	3e (73%)	—
(5) 1b (R ¹ = Me)	2c (R ² = <i>n</i> -Bu)	3f (69%)	—
(6) 1c (R ¹ = CH ₂ OBn)	2b (R ² = Me)	3g (75%)	—
(7) 1c (R ¹ = CH ₂ OBn)	2c (R ² = <i>n</i> -Bu)	3h (78%)	—
(8) 1d (R ¹ = 3-1)	2b (R ² = Me)	3i (41%)	4i (18%)

^a [aldehyde] = 0.17 M, alcohol (2.5 equiv), 1,4-dioxane, PtCl₂ (10 mol %), CO (1 atm), 80 °C, 2 h. ^b Product yields are reported after silica column chromatography.

elucidation of product frameworks relies on an X-ray structure of one diastereomer of compound **6h**. The results in entries 1–2 and 5–6 reveal that the electron-rich phenylethynyl substituents of enynals **5b–d** gave better yields (67–80%) of annulation products **6b**, **6e**, and **6f** as compared to **6a** (61% yield) given from parent species **5a**. This new catalytic annulation is extendible to allylic alcohols **2c** and **2a** (R = *n*-Bu, Ph; entries 3–4), giving tricyclic ketones **6c–6d** and **6g** in reasonable yields. The use of acyclic enynal **5f** gave bicyclic ketone **6h** in 42% yield. This new annulation provides a rapid construction of the tricyclic cores of natural compounds represented by sapogenol,^{9a} abruslactone,^{9b} diosbulbin-B,^{9c} and lungshengenins D.^{9d}

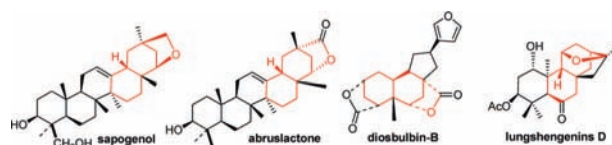
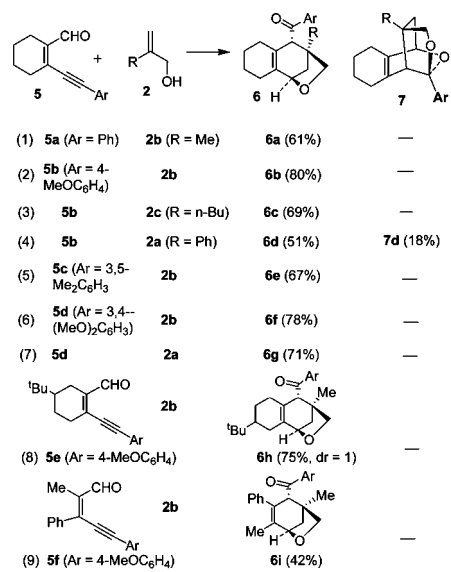
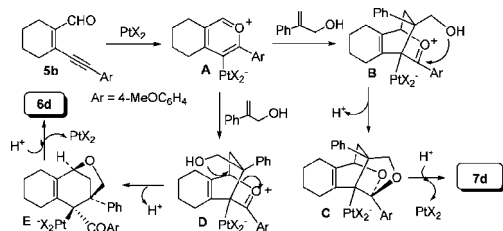


Table 3. Platinum-Catalyzed Annulation of Enynals with Allylic Alcohols

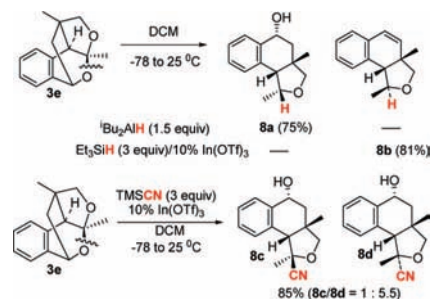
^a [aldehyde] = 0.17 M, alcohol (2.5 equiv), toluene, PtCl₂ (10 mol %), AgOTf (20 mol %), 23 °C, 16 h. ^b Product yields are reported after silica column chromatography.

Scheme 1

The isolation of ketal species **7d** enabled us to elucidate the formation mechanism of tricyclic oxacyclic species **6d**, generated from a subtle annulation of enynal **5b** with 2-phenylallylic alcohol. As shown in Scheme 1, a minor proportion of pyrilium **A** undergoes [4 + 2]-cycloaddition with the *si*-face of allylic alcohol, giving intermediate **B** with a tethered alcohol to trap this species through formation of ketal species **C** that ultimately gives observed ketal **7d**. We envisage that a major proportion of pyrilium species **A** undergoes cycloaddition with the *re* face of allylic alcohol to give adduct **D**, of which the tethered alcohol facilitates dissociation of the ketone via an intramolecular S_N2 attack, producing species **E** with controlled stereochemistry. With this reaction model, we conclude that the distinct pathways in Tables 2 and 3 stem from separate diastereofacial faces in the cycloadditions of allylic alcohols with benzopyrilium and pyrilium intermediates, arising from 2-alkynylbenzaldehydes **1a–d** and enynals **5a–f**, respectively.

Scheme 2 shows selected examples for stereocontrolled cleavage of the ketal functionality of species **3e**. Treatment of this ketal with ^tBu₂AlH¹⁰ (1.5 equiv) gave fused tricyclic tetrahydrofuran **8a** in 75% yield. With our own efforts, we found that Et₃SiH (3 equiv)/10% In(OTf)₃ and TMSCN (3 equiv)/10% In(OTf)₃ effected cleavage of the ketal ring of species **3a**, with retention of stereochemistry, giving **8b** (81%) and **8d** (ca. 72%) exclusively.

Before this work, the [4 + 2]-cycloadducts from benzopyriliums (or pyriliums) and alkenes are merely a hypothetical intermediate without actual use. Our new strategy to intercept [4 + 2]-cycloadd-

Scheme 2

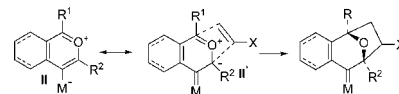
ducts involves the use of 2-substituted allylic alcohols. For nonaromatic enynals, we obtained distinct tricyclic oxacyclic ketones due to occurrence of a new annulation reaction. The values of such cycloadditions and annulations are reflected by their high diastereoselectivities and chemoselectivities. New approaches to intercept [4 + 2]-cycloadducts from benzopyriliums and alkynes are under current investigations.¹¹

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Supporting Information Available: Experimental procedures, X-ray crystallographic data of compounds **3i** and **6h**, NMR spectra, and spectra data for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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- (7) Crystallographic data of compounds **3i** and **6h** are provided in Supporting Information.
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